

Potential impacts of new underground land use practices on groundwater quality in the context of the energy transition

Philippe Gombert*, Elodie Lacroix*, Stéphane Lafortune*,
Angélique Poulain**

** Ineris, France*

*** University of Mons, Belgium*

10/09/2019



OUTLINE

- ① The underground storages in the context of the energy transition**
- ② The new energy products to be stored underground in the near future**
- ③ The potential geochemical risks of these emerging technologies**
- ④ Conclusion**



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The underground storages in the context of the energy transition

The 2015' French law relative to the energy transition and its implications

Main objectives of the French law relative to the energy transition (by 2050):

- / **Divide by 2** the global energy consumption
 - / **Divide by 4** the greenhouse gas emissions
- } **Increase x 4** the Renewable Energy production

Main problems:

- / **Hydropower** is the main renewable energy source (in F) but there are **no new potential sites**
- / **Eolian** and **solar** energy are **intermittent** ⇒ **Storage solutions** are needed

Main advantages of underground storage:

- / **Good feedback** (>600 hydrocarbon storage sites in the world since 1915)
- / **More safe** than surface storage
- / **Less footprint**

Main disadvantages of underground storage:

- / Difficulty of underground **access** ⇒ **higher costs** (CAPEX)
- / Risks of **underground perturbations** ⇒ geomechanical, thermal and hydrogeochemical.

Different Underground Storage Types and Stored Products (in France)

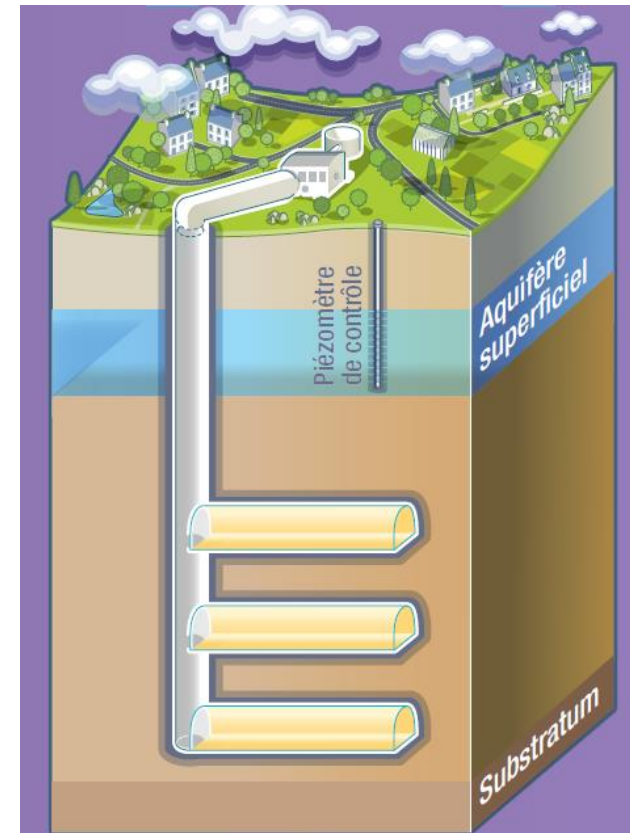
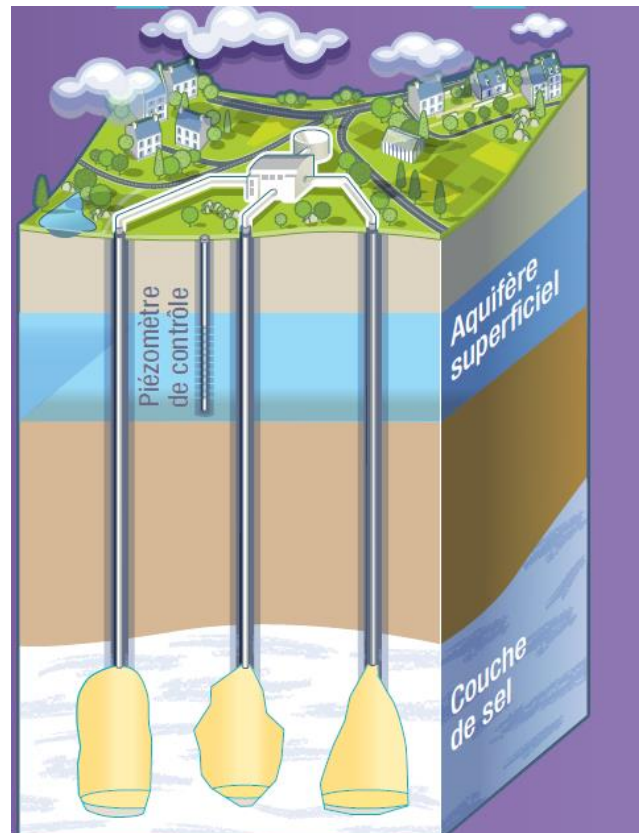
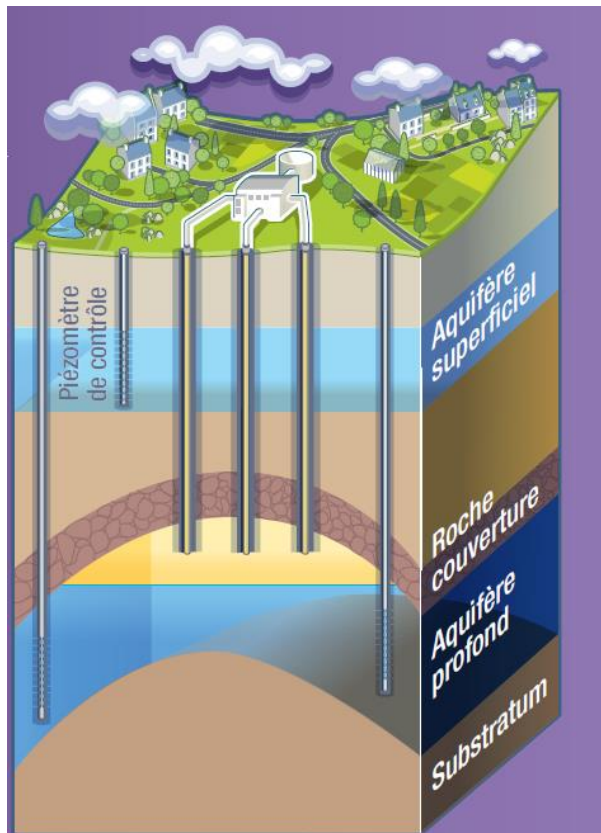
Three underground storage types:

- / in (deep) **aquifers** (13)
- / in **saline dissolution cavities** (78)
- / In **mined cavities** (9)

Four stored energetic products:

- / Natural gas (11 Gm³)
- / Natural gas (2 Gm³), hydrocarbons (14 Mm³)
- / Mainly LPG (0.5 Mm³)

Fossil fuels.





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**The new energy products to be stored
underground in the near future:**

- Hydrogen
- Compressed Air
- Water (Warm or Cold)

The Underground « Green » Gas Storages: 1) H₂ and 2) Compressed Air

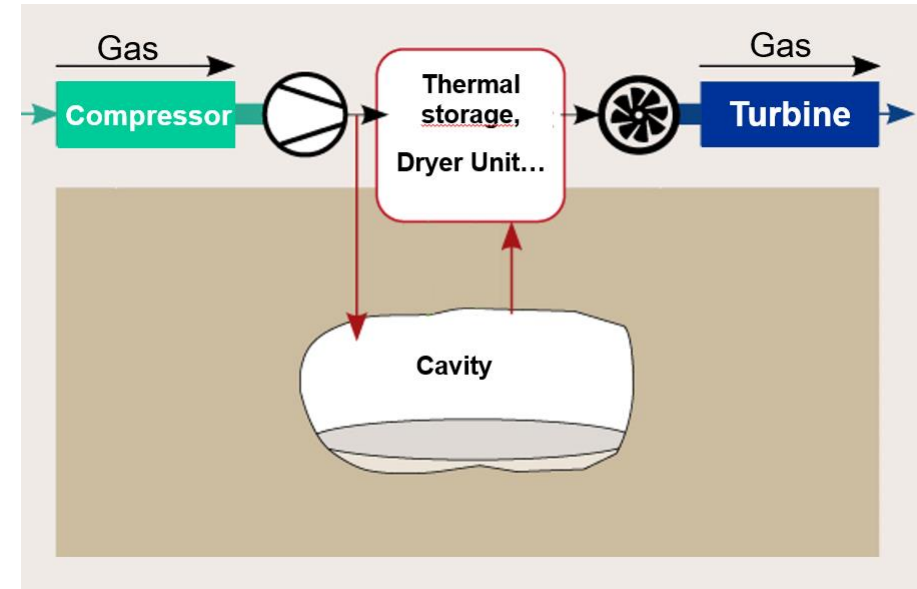
HUS = Hydrogen Underground Storage – CAES = Compressed Air Energy Storage

Principle:

- / To produce H₂, then compress it and store underground
⇒ to burn it to produce electricity (or to use it as eco-fuel)
- / To compress air, then store it in underground reservoir
⇒ to relax it through a turbine to produce electricity

Achievements in the world:

- / 4 Underground Hydrogen storages (3 in USA, 1 in UK)
- / 2 Underground Compressed Air Energy Storages (D, USA)



Potential geochemical impacts in case of leakage:

- / H₂ is a **reductor** agent ⇒ potential **reduction** of oxygenated water from shallow aquifers
- / Air is an **oxydant** agent ⇒ potential **oxydation** of anoxic water from deep aquifers.

3) The Underground Pumped Hydro-Energy Storage (PHES)

Principle:

/ To use electricity to transfer water from a lower to an upper reservoir $\Rightarrow$ To turbine water downwards

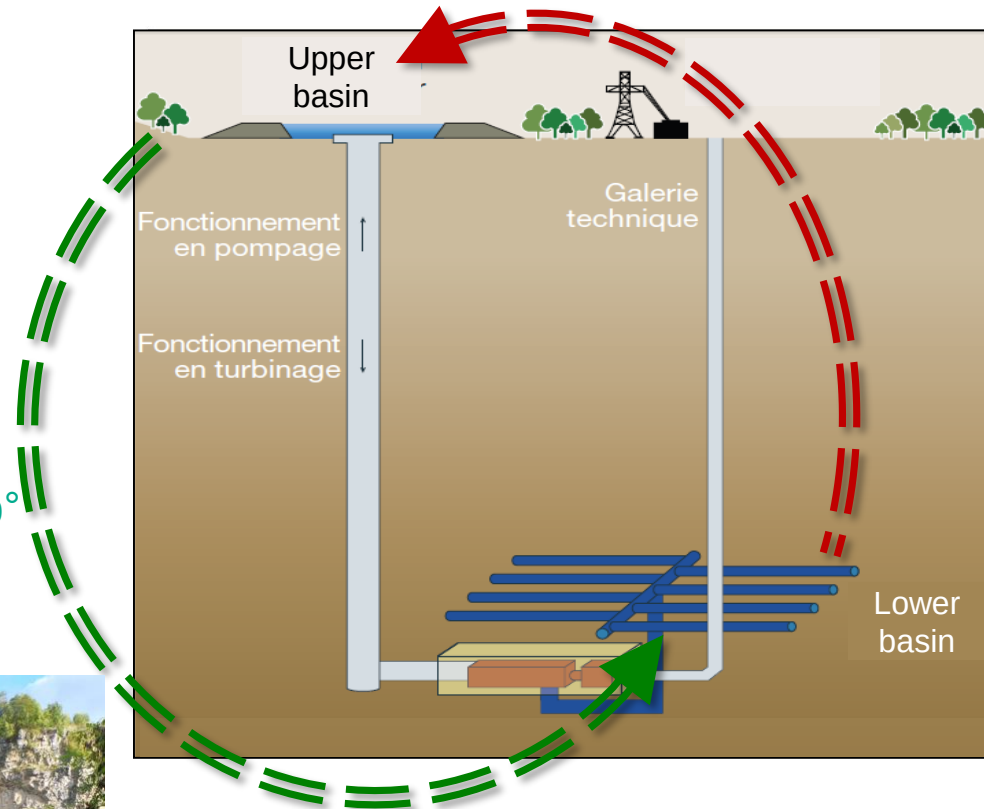
- Q-PHES: the lower reservoir is an abandoned **Quarry** basin

Achievements in the world:

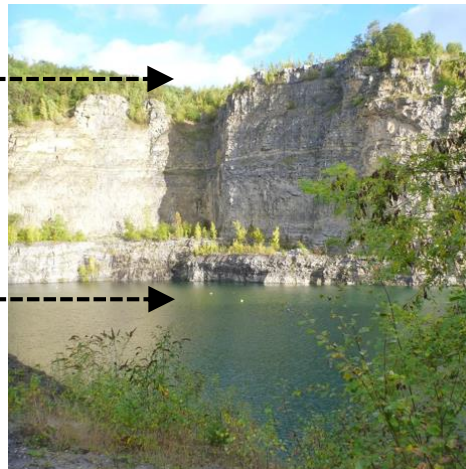
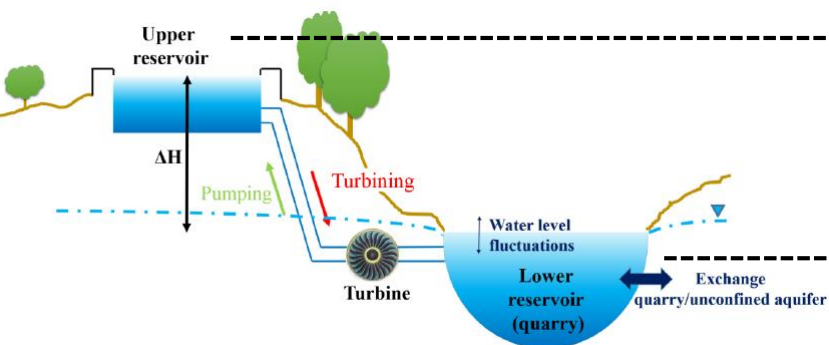
- / M-PHES: 1 project in a former coal mine in D
- / Q-PHES: None but several hundreds of potential sites

Potential geochemical impacts:

- / M-PHES: oxygenation of anoxic mine water $\Rightarrow$ Fe/Mn pp°
- / Q-PHES: oxygenation of groundwater $\Rightarrow$ $CaCO_3$ pp° .



**Most important energy storage technology
~400 surface PHES but none underground**



4) The Underground Thermal Energy Storage (A-TES and C-TES)

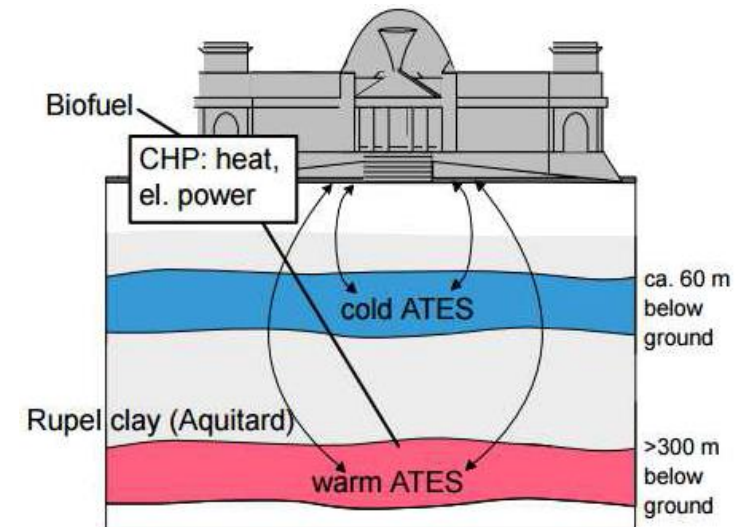
A-TES = Aquifer Thermal Energy Storage – C-TES = Cavern Thermal Energy Storage

Principle:

- / To store excess heat produced in Summer
- / To reuse it during Winter (and vice versa with cold in Winter)

Achievements in the world:

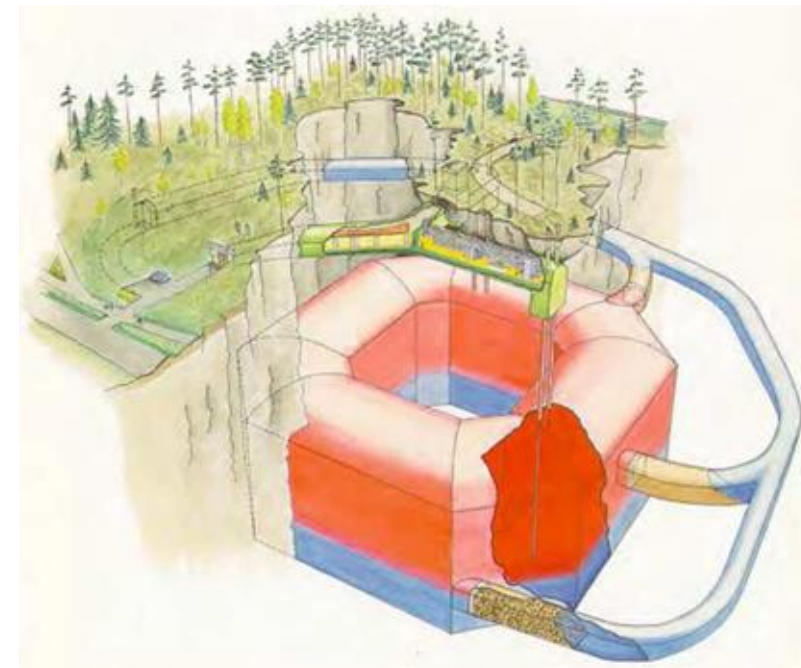
- / A-TES (Aquifer): > a **thousand** in the world (NL, D, USA)
- / C-TES (Cavern): < a **dozen** in the world (Scandinavia)



ATES at Berlin (D) ↑

Potential geochemical impacts:

- / Dissolution/Precipitation processes due to fluctuations of water temperature (min. 5°C ⇒ max. 95°C).



CTES at Lyckebo (S) →
(104,000 m³ of water at 60-90°C)



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**The potential geochemical risks of
these emerging technologies:**

- Hydrogen storage
- Quarry-PHES

Potential Impacts of Hydrogen Underground Storages on Shallow Aquifers

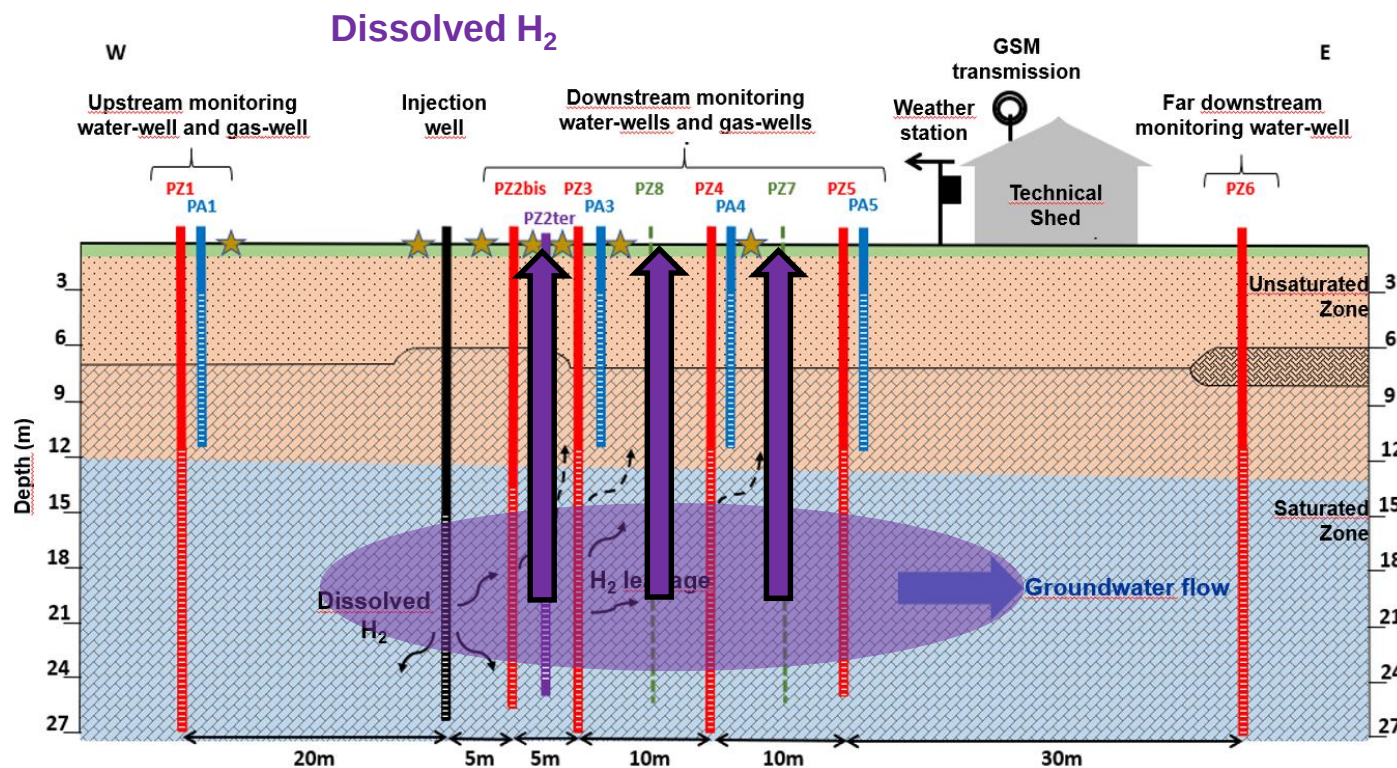
ROSTOCK'H: Risks and Opportunities of the Hydrogen geological Storage in salt cavities in France and Europe

Objectives of the project:

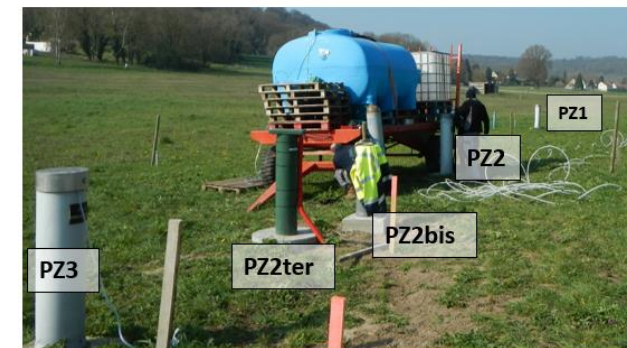
- ① To develop in-situ H_2 monitoring tools in SZ and UZ from leakage experiment at an experimental site

Bibliographic synthesis + PHREEQ-C model:

- / Abiotic reduction of **sulphates**: $4 H_2 + SO_4^{2-} + 2 H^+ = H_2S + 4 H_2O$
- / Abiotic reduction of **nitrates** with **catalyzers**:
 - Steel and stainless steel (borehole/well casings): $NO_3^- + 4 H_2 + 2 H^+ = NH_4^+ + 3 H_2O$
 - Pd and Cu: $2 NO_3^- + 5 H_2 = N_2 + 2 OH^- + 4 H_2O$



Previous experiments (CO_2 , He)



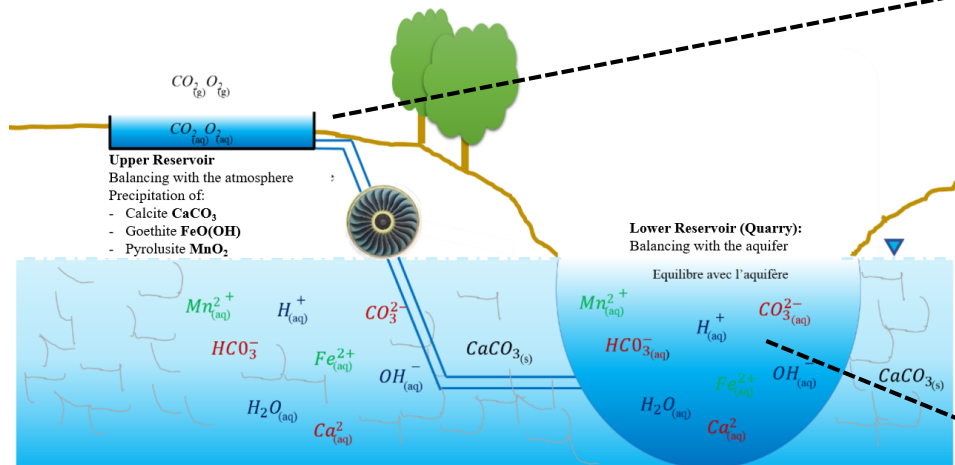
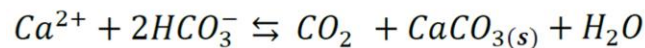
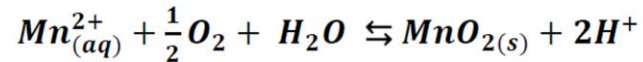
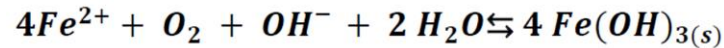
Potential Impact of Q-PHES on an Adjacent Aquifer

(from Angélique Poulain's PhD Thesis, University of Mons, B, 2018)

Simulation of groundwater flow and solute-transport by PHAST (daily cycles):

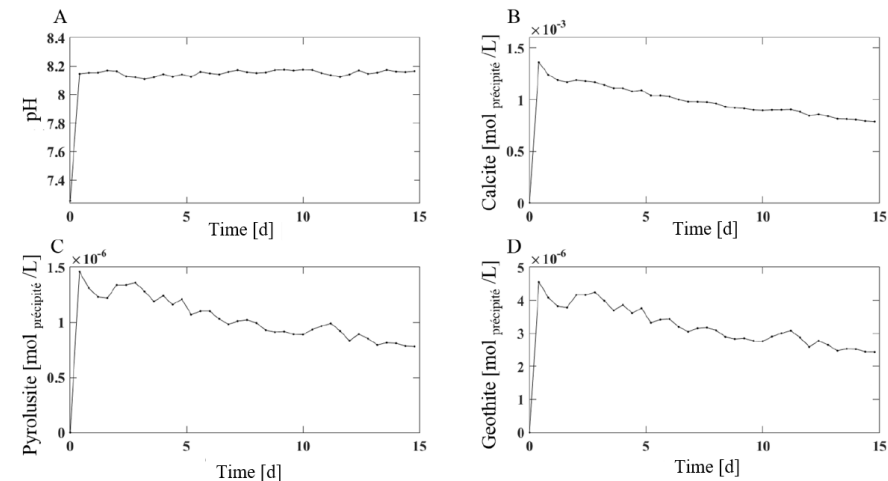
/ In the upper basin: $[O_2] \uparrow + [CO_2] \downarrow \Rightarrow$ precipitation of calcite, goethite and pyrolusite, +1 pH unit

Daily cycle = 1 Mm³ $\Rightarrow$
140 t of calcite, 0.4 t of goethite and 0.1 t of pyrolusite precipitated in the upper basin !



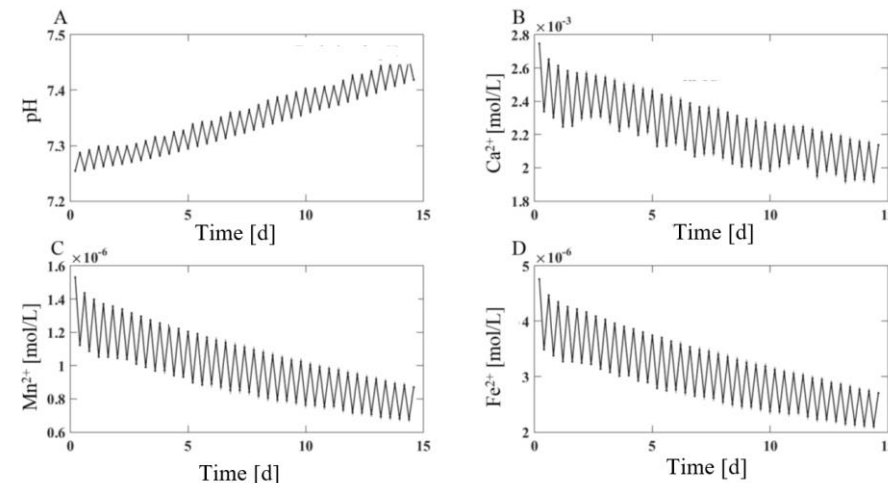
Different chemical elements involved in an open aquifer system

/ In the quarry and the aquifer $\Rightarrow$ cyclic qualitative variations induced by daily cycles between upper basin and aquifer values: $[Ca] \downarrow$, $[Mg] \downarrow$, $[Mn] \downarrow$, $[Fe] \downarrow$, $pH \uparrow$ (+0.2 unit).



In the upper basin $\uparrow$
Evolution of hydrogeochemical variables during 2 weeks

In the lower basin $\downarrow$

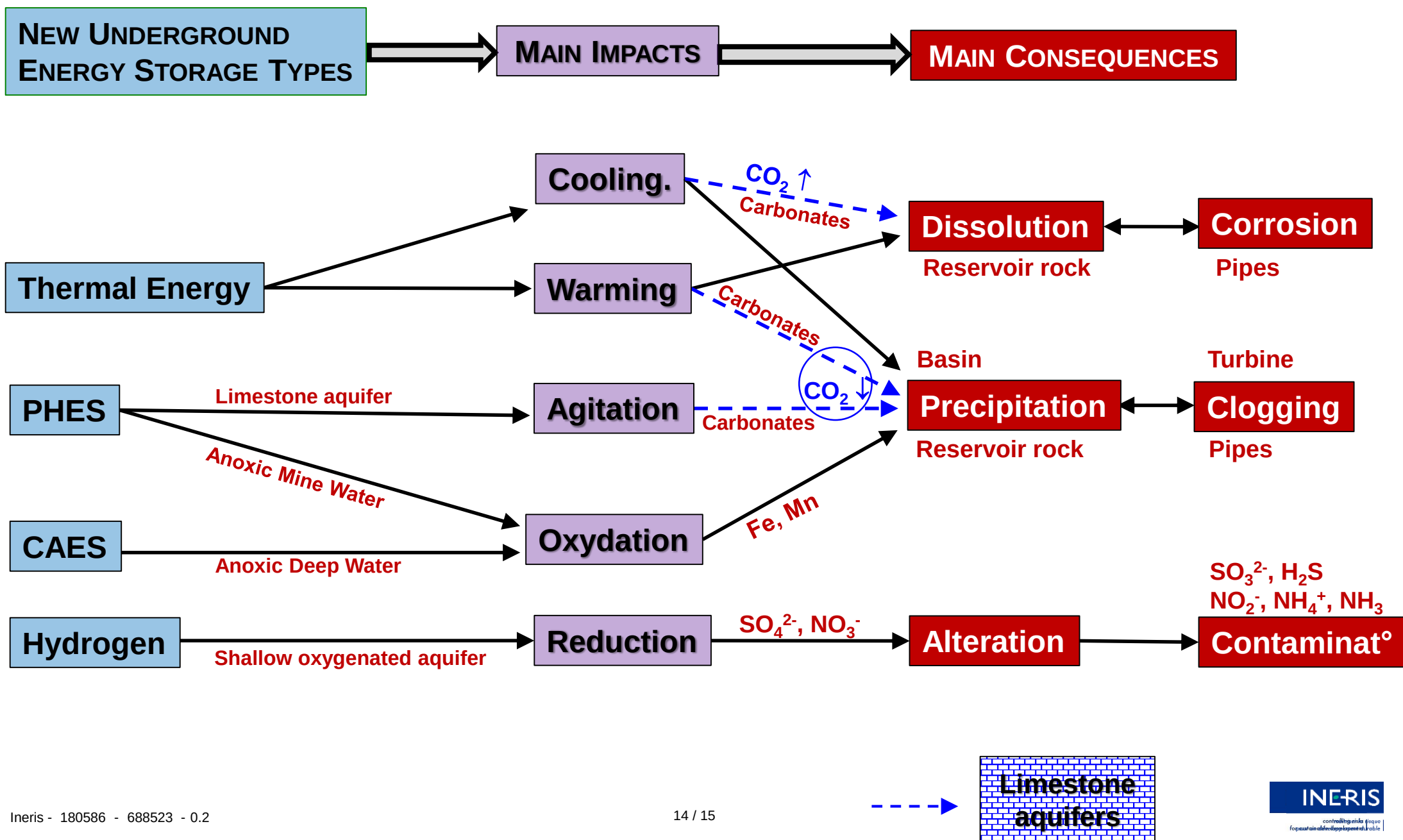




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Conclusion

Geochemical Risks devoted to New Underground Energy Storages



Thank you for your attention
Merci de votre attention
Dank U voor uw aandacht

Data come from a
report available
(in French) at:
www.ineris.fr

