

Spatial distribution and temporal evolution of chlorinated solvents in a drinking water ressource

Justine Criquet¹, Mélinda George¹, Aurélie Bouvet², Jean-Noël Ottenwaelder², Miléna Walasek^{1,3}, Lise Cary³, Emmanuel Bugner² and Gabriel Billon¹

¹ Université de Lille - LASIR Laboratory

² Métropole Européenne de Lille

³ BRGM Hauts-de-France



Context: Lille Metropolis (France)

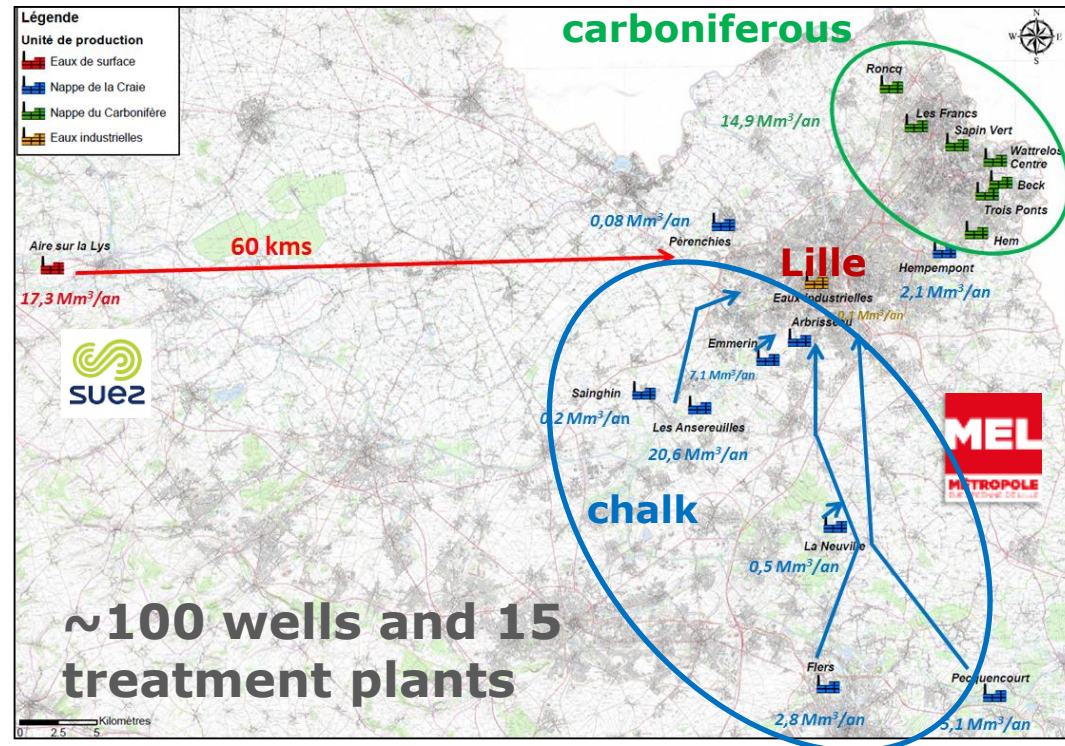
- 1.2 million inhabitants (4th in France)
- In 2018 : production of 70.8 million m³ of drinking water (200 000 m³/d)



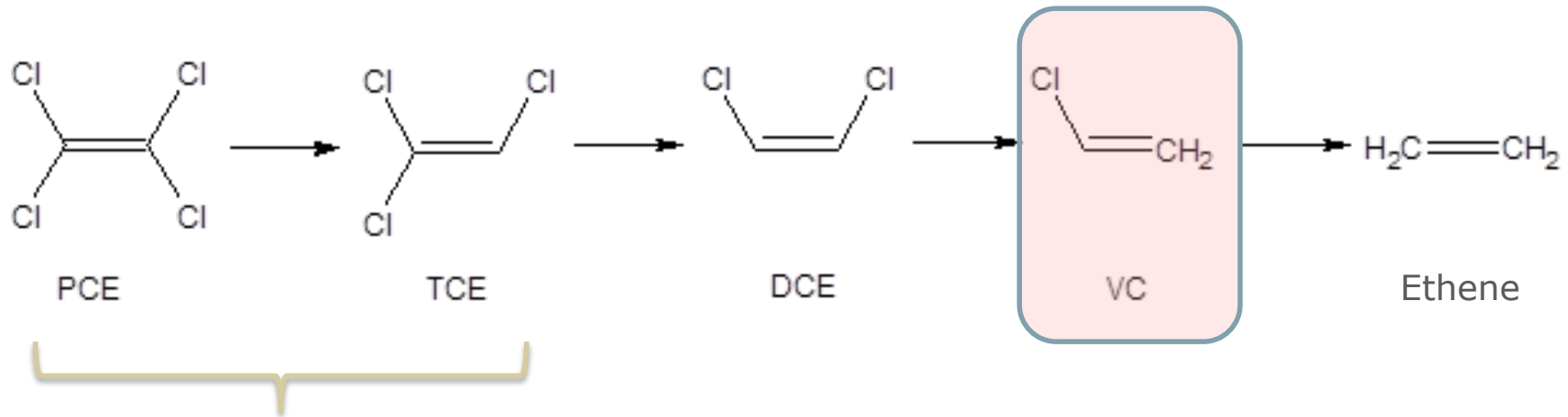
Surface water (Lys River – 24 %)

Groundwaters (76%)

- « Available water »: 120%
- 40% of the ressource is impacted by chlorinated solvents from former industries



Simplified scheme of tetrachloroethylene degradation:



*EU
Drinking
water
regulation*

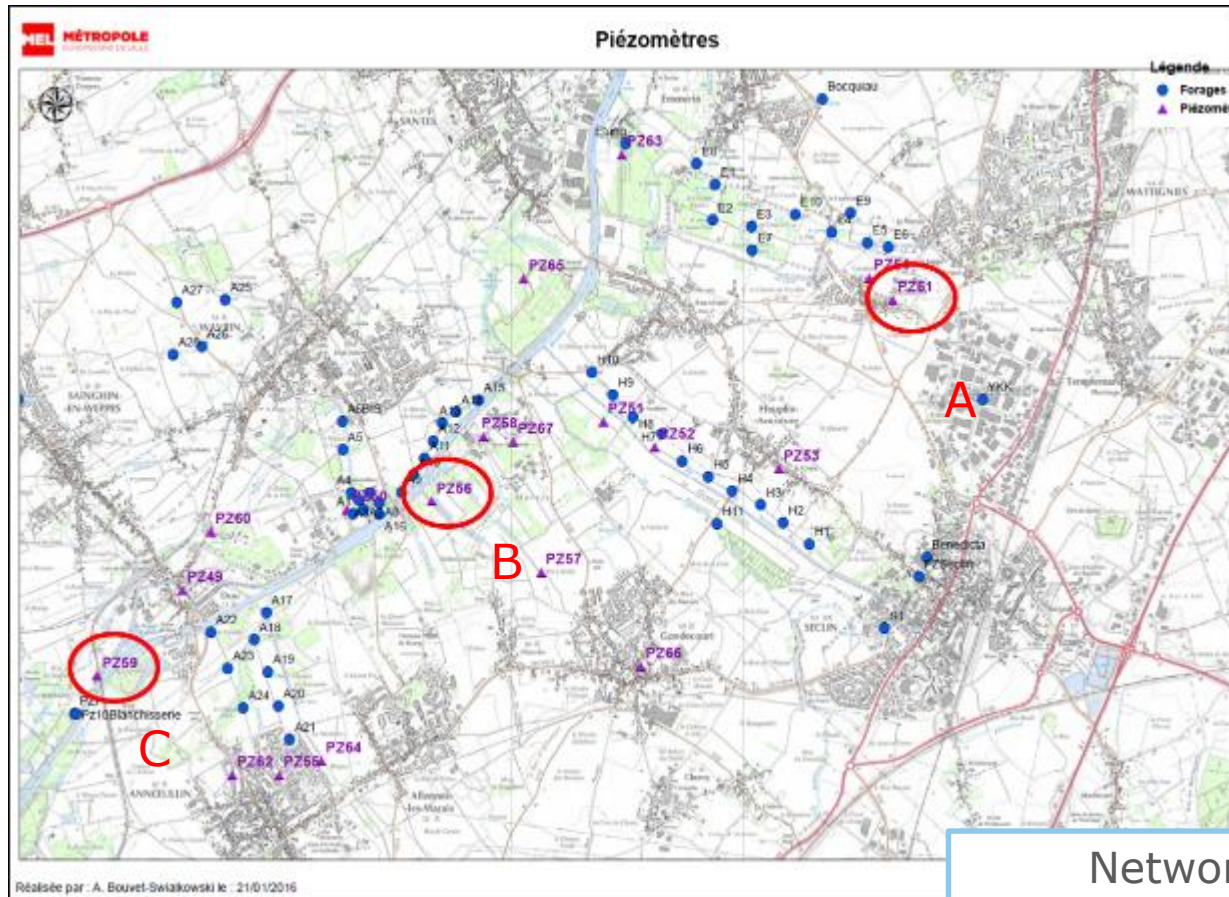
Σ tetrachloroethylene +
trichloroethylene
 $< 10 \mu\text{g L}^{-1}$
respected by dilution

Vinyl chloride
 $< 0.5 \mu\text{g L}^{-1}$

Is this water resource
sustainable ?

- Presence of multiple sources of PCE, TCE, DCE,...
- DCE has been detected at concentrations $> 350 \mu\text{g L}^{-1}$ in this aquifer

Localization



- Production wells
- ▲ Monitoring piezometers dedicated to VOC

9 VOCH analysed:

tetrachloroethylene
trichloroethylene
Cis-1,2-dichloroethylene
Trans-1,2-dichloroethylene
1,1-dichloroethylene
vinyl chloride
dichloroethane
trichloroethane

Network of 19 piezometers
dedicated to monitoring

diameter: 168.3 mm – stainless still
(to avoid vinyl chloride leaching)

Equilibrium based passive sampler (no accumulation)

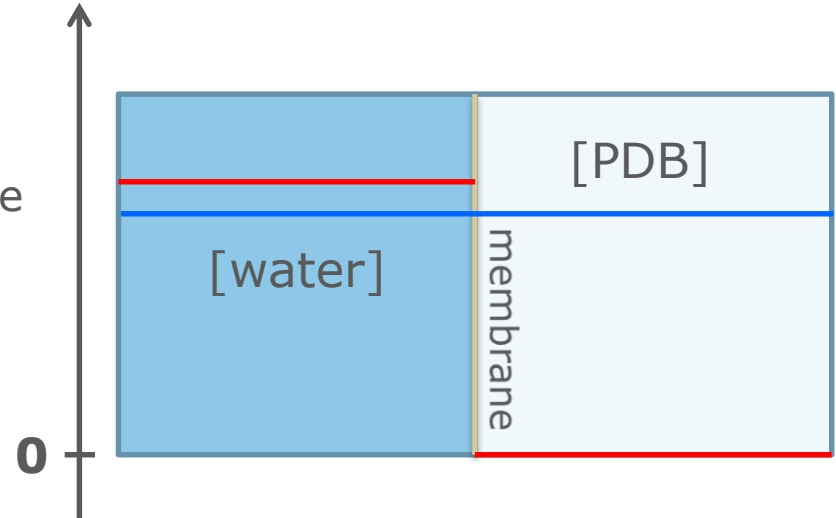


PDB = Polyethylene Diffusion Bag

- Membrane: low density polyethylene
- Filled with ultrapure water
- Thickness: 100 μm
- 90 cm high,
- 4 cm large
- Volume: 300 mL

4 campaigns /year
PDB at different depths

[Conc]



$t = 0$

$t = 2 \text{ weeks}$

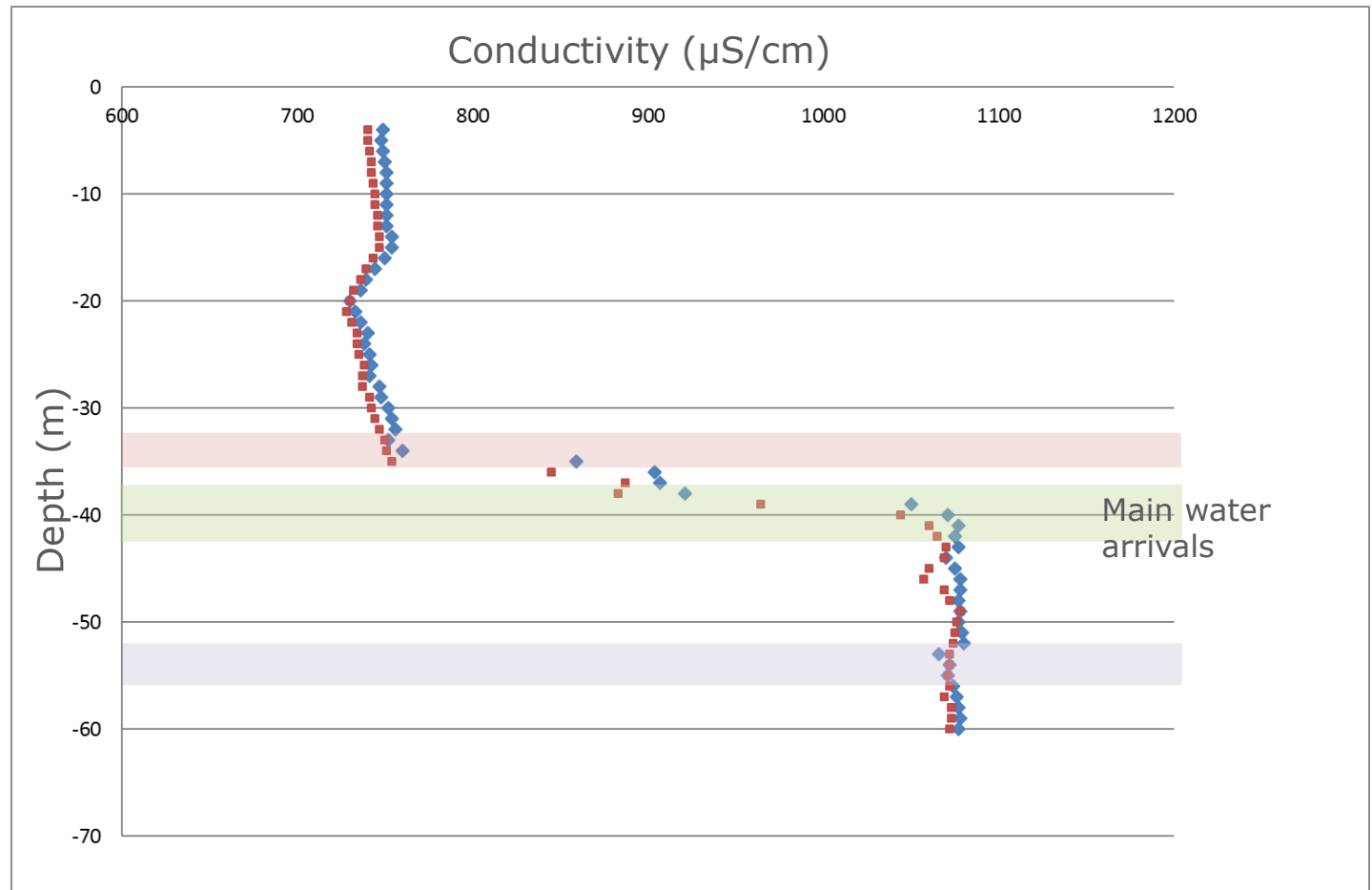
$[\text{water}] = C_0$
 $[\text{PDB}] = 0$

$[\text{water}] = [\text{PDB}]$

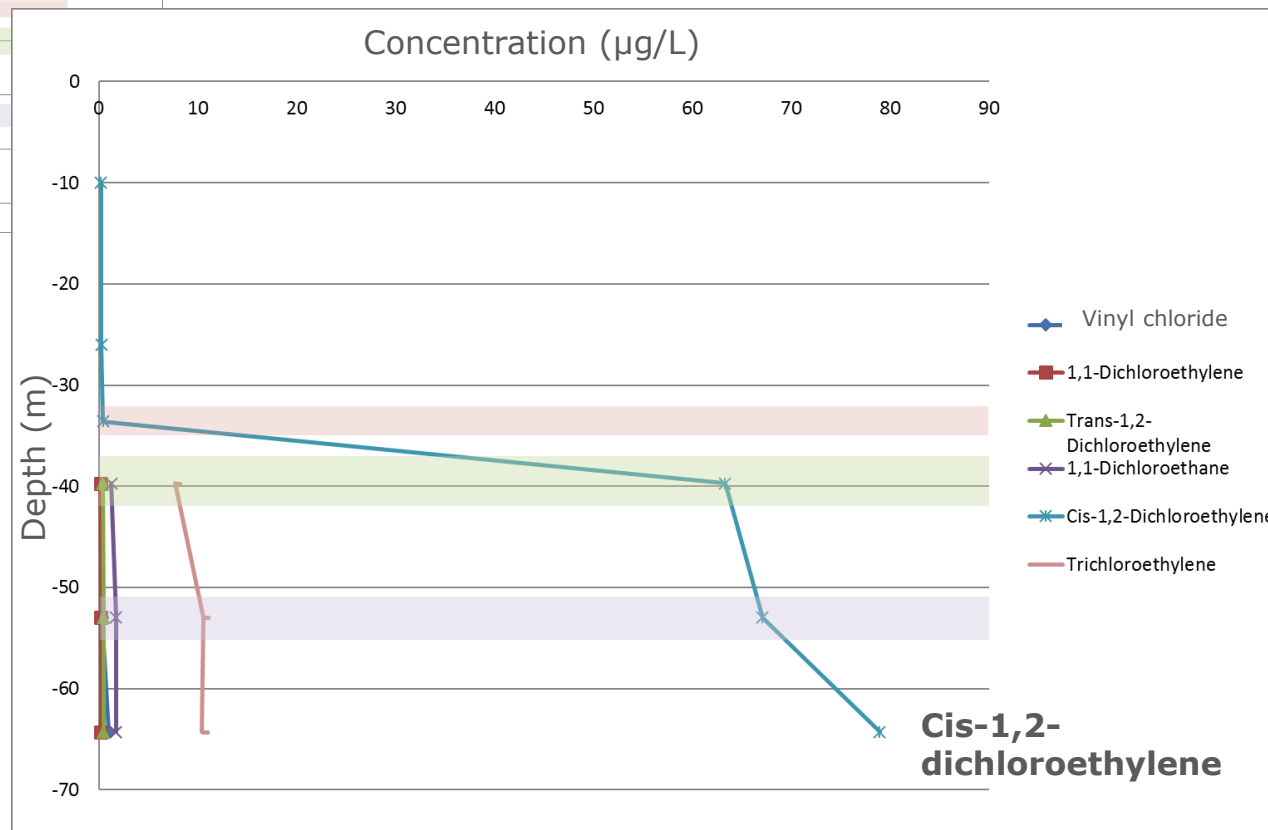
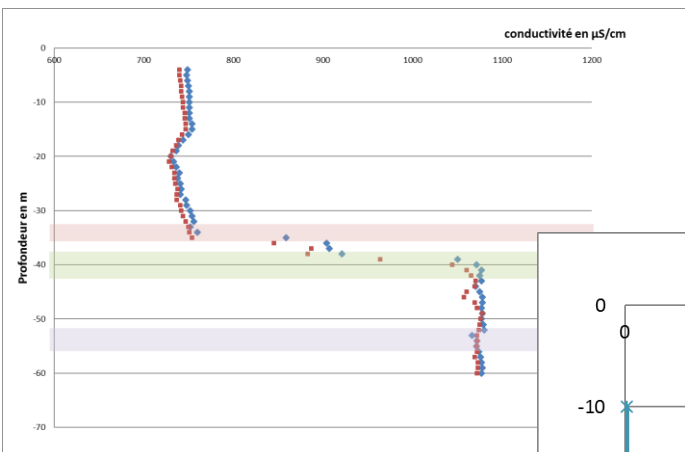
Extra characterization in function of depth:

- dissolved oxygen, conductivity, t°
- anions, cations, trace metals and pesticides

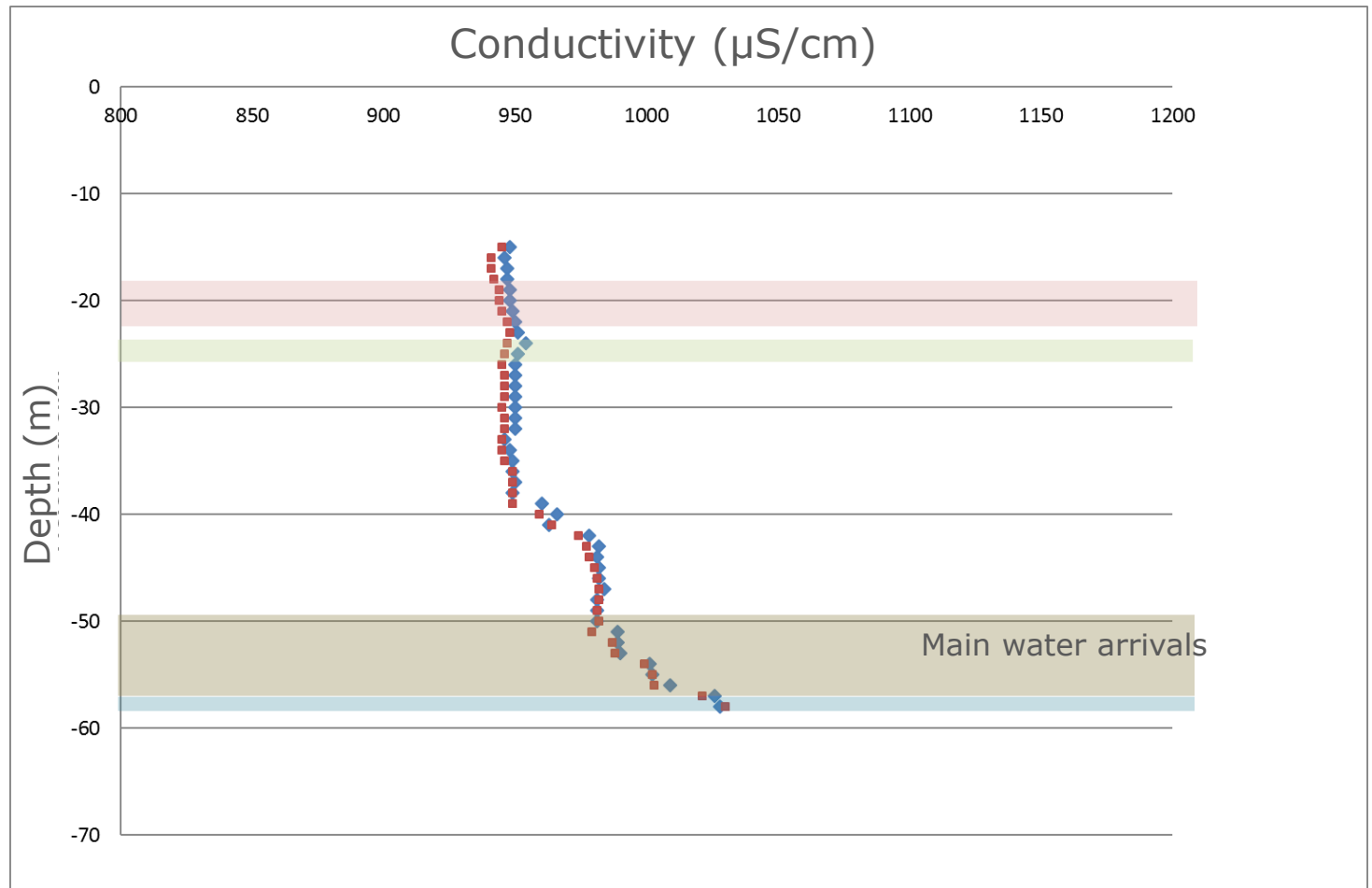
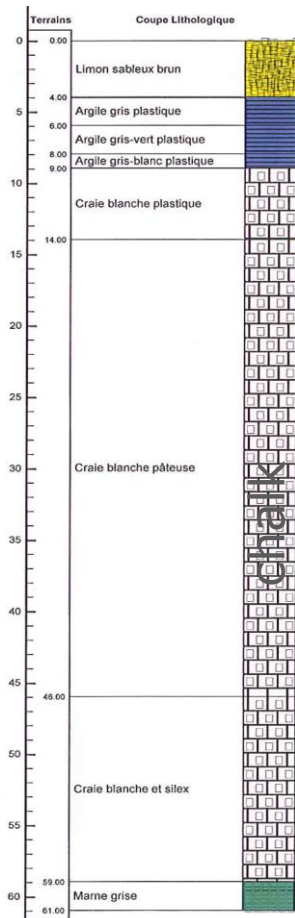
Poster n°105
Billon et al.

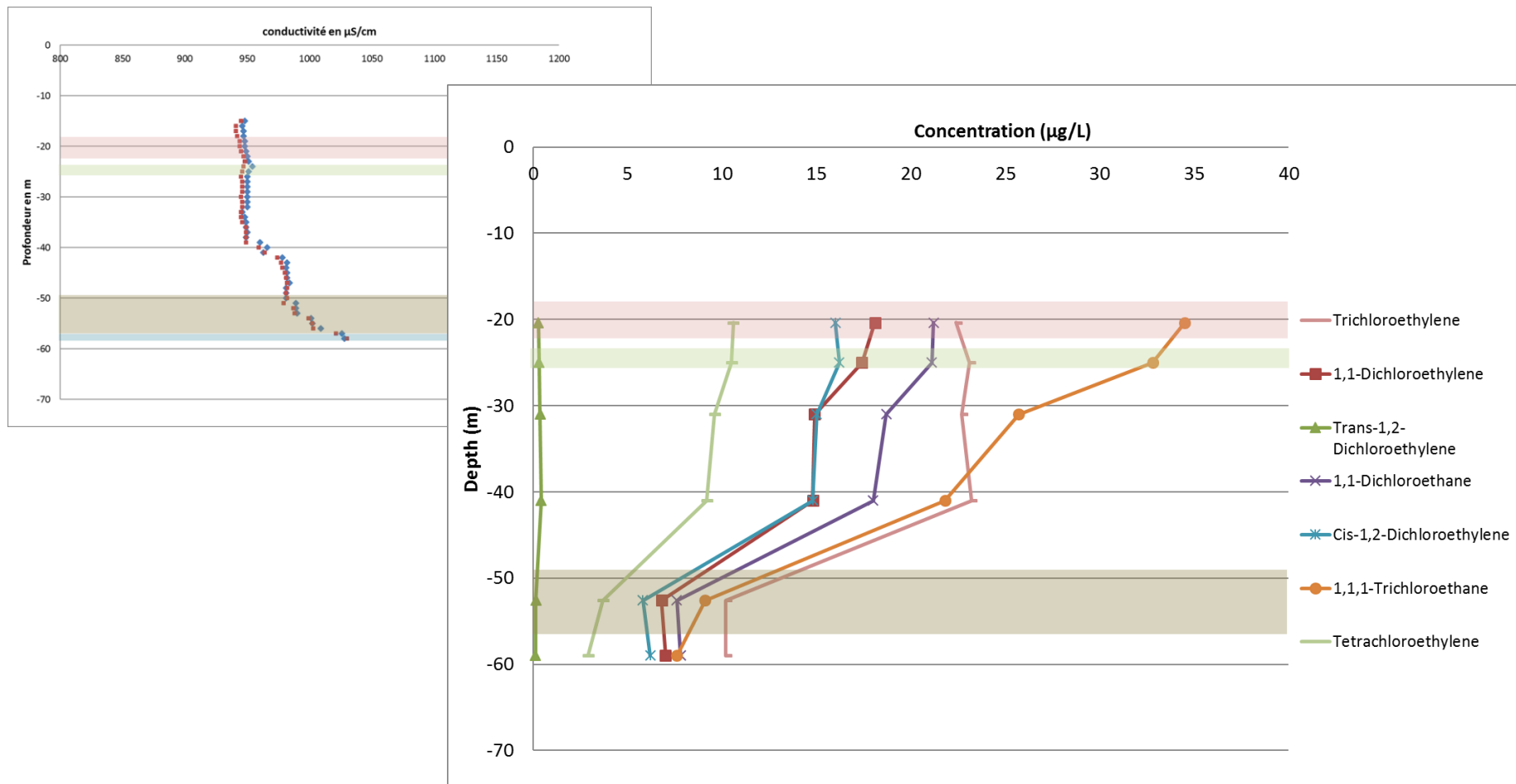


- Sudden change of water characteristics:
sampling with dialysis cells: $[\text{Ca}^{2+}]$ from 100 to 180 mg/L
 $[\text{Na}^+]$ from 60 to 80 mg/L
- No mixing of the water column with classical sampling



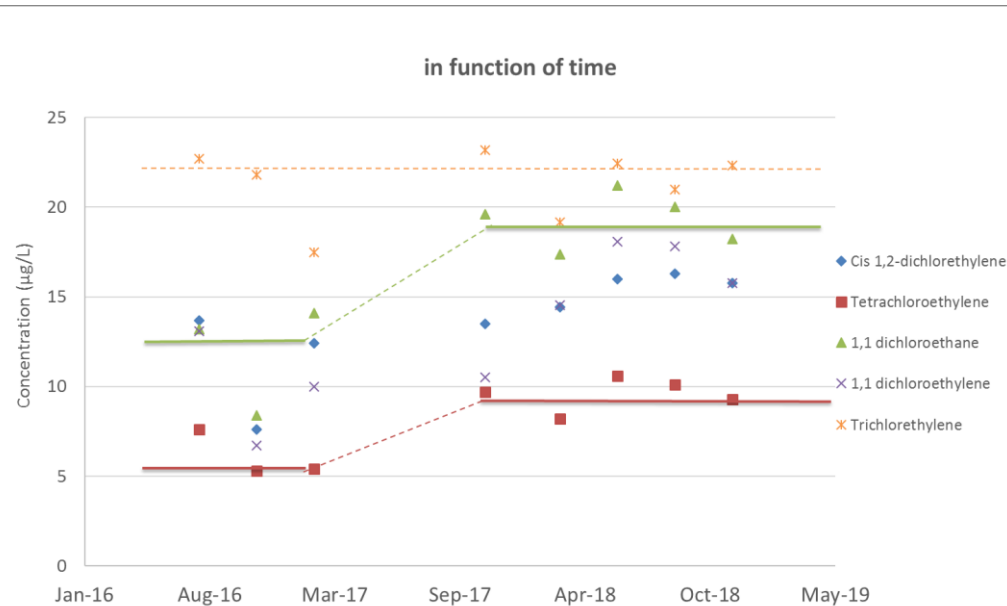
- High concentration of cis-1,2-dichloroethylene (80 µg/L)
- Vinyl chloride from 0.3 to 1 µg/L





7 / 9 compounds detected
 Similar behaviour for all compounds
 Lower with depth

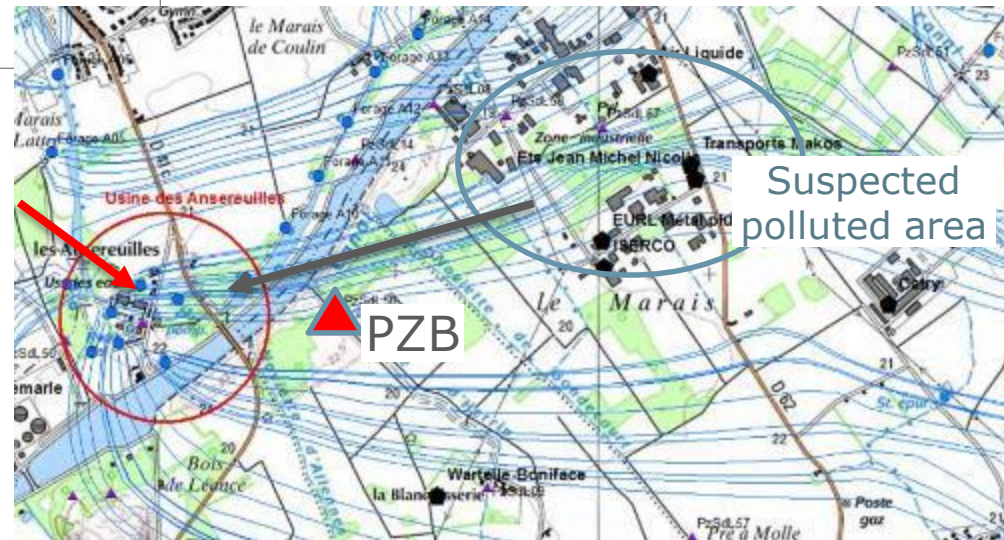
PZ B: temporal evolution



- Increase of chlorinated solvent levels since 2016
- Except for trichloroethylene
- Existing correlation between the concentrations of the different substances except trichloroethylene

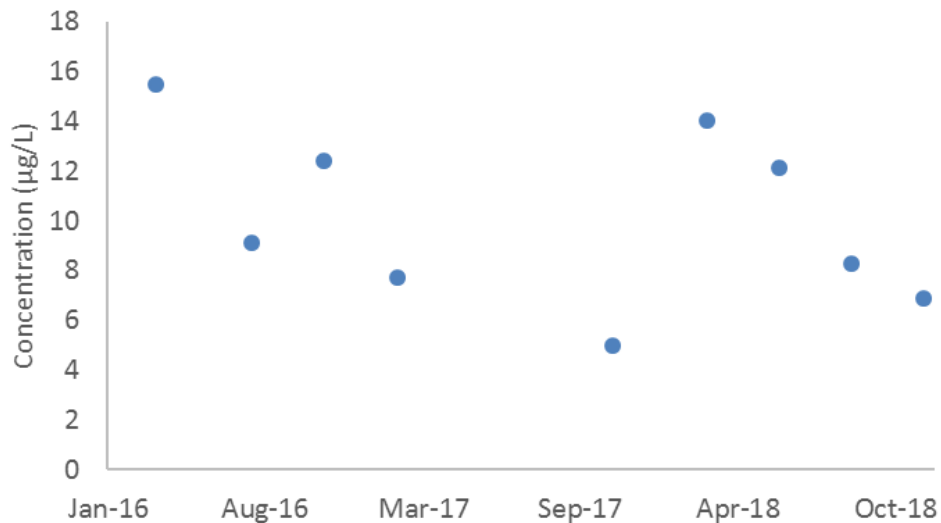
- Increase of the water demand in this area
- Higher transfer from a suspected polluted area

Water
treatment
plant



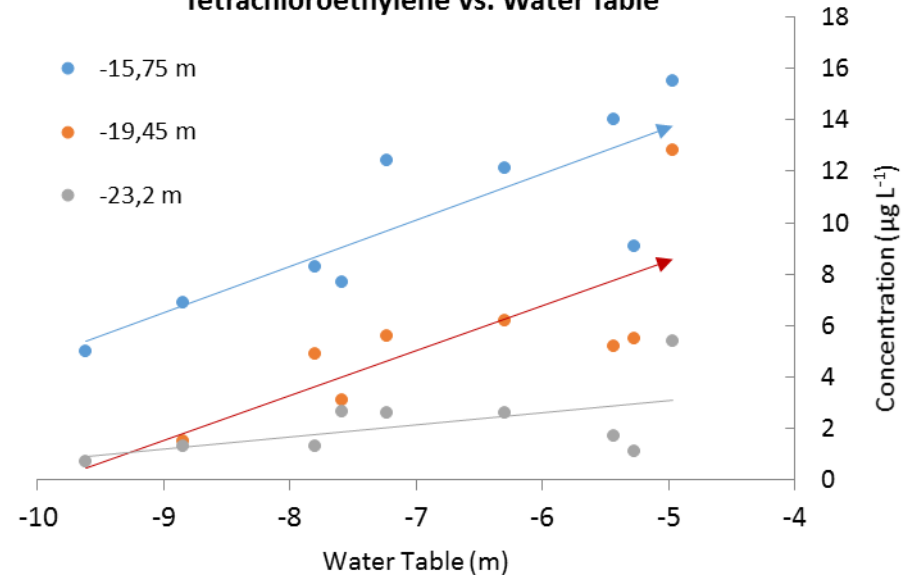
PZ C: evolution with water table

Tetrachloroethylene vs. Time



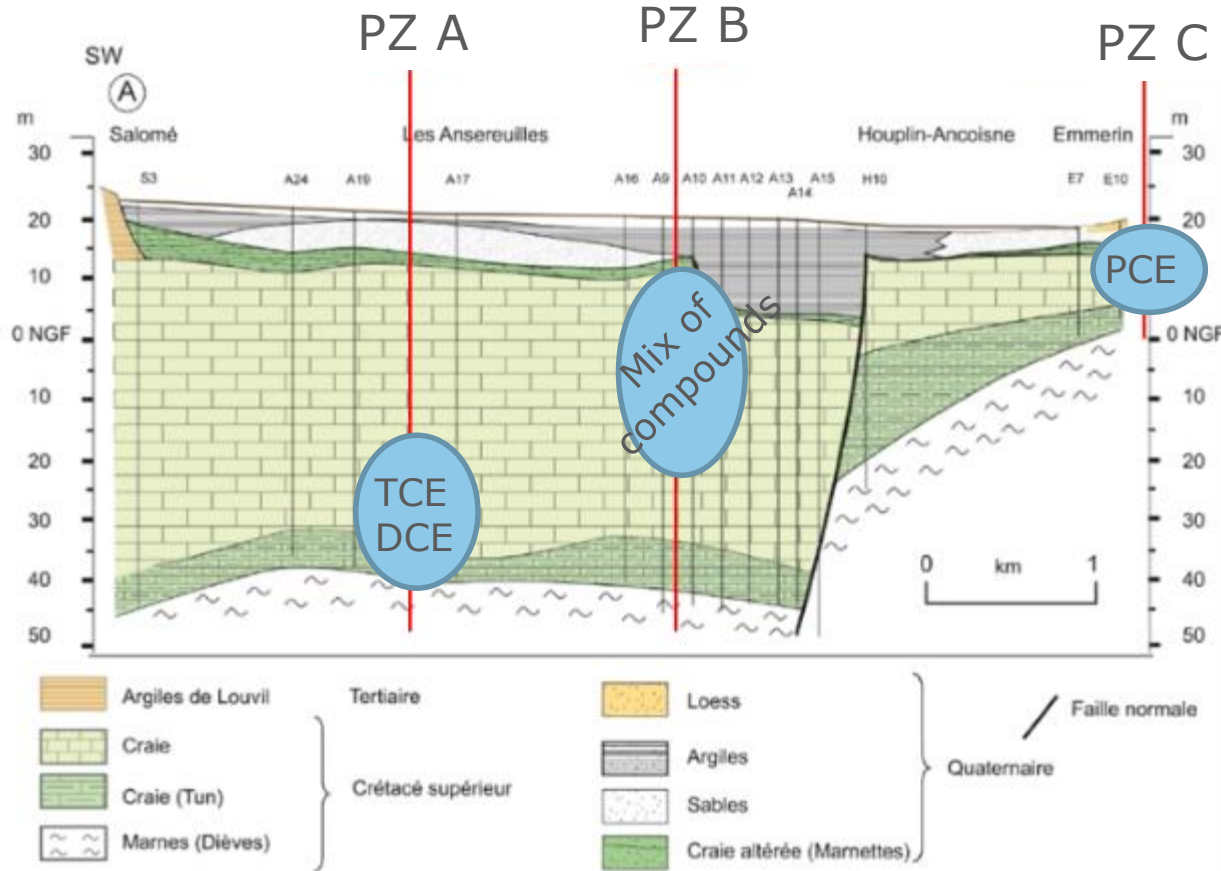
- No particular evolution of concentrations in function of time
- No clear seasonnal effect

Tetrachloroethylene vs. Water Table



- The concentrations of chlorinated solvent increase with the increase of water table
- Remobilization of chlorinated solvents during aquifer recharge

Conclusion



- Passive sampler: useful tool to better characterize the aquifer
- Highly heterogeneous pollution / multiple sources
- Further work needed to predict the evolution of the aquifer

Thank you



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